

(Still) More Naming Practice

Write the names of the following chemical compounds:

- 1) BBr_3 _____
- 2) CaSO_4 _____
- 3) C_2Br_6 _____
- 4) $\text{Cr}(\text{CO}_3)_3$ _____
- 5) Ag_3P _____
- 6) IO_2 _____
- 7) VO_2 _____
- 8) PbS _____
- 9) CH_4 _____
- 10) N_2O_3 _____

Write the formulas of the following chemical compounds:

- 11) tetraphosphorus triselenide _____
- 12) potassium acetate _____
- 13) iron (II) phosphide _____
- 14) disilicon hexabromide _____
- 15) titanium (IV) nitrate _____
- 16) diselenium diiodide _____
- 17) copper (I) phosphate _____
- 18) gallium oxide _____
- 19) tetrasulfur dinitride _____
- 20) phosphorus heptafluoride _____

(Still) More Naming Practice - Answers

Write the names of the following chemical compounds:

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|-----|----------------------------|--------------------------------|
| 1) | BBr_3 | boron tribromide |
| 2) | CaSO_4 | calcium sulfate |
| 3) | C_2Br_6 | dicarbon hexabromide |
| 4) | $\text{Cr}(\text{CO}_3)_3$ | chromium (VI) carbonate |
| 5) | Ag_3P | silver phosphide |
| 6) | IO_2 | iodine dioxide |
| 7) | VO_2 | vanadium (IV) oxide |
| 8) | PbS | lead (II) sulfide |
| 9) | CH_4 | methane |
| 10) | N_2O_3 | dinitrogen trioxide |

Write the formulas of the following chemical compounds:

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|-----|-----------------------------|---|
| 11) | tetraphosphorus triselenide | P_4Se_3 |
| 12) | potassium acetate | $\text{KC}_2\text{H}_3\text{O}_2$ |
| 13) | iron (II) phosphide | Fe_3P_2 |
| 14) | disilicon hexabromide | Si_2Br_6 |
| 15) | titanium (IV) nitrate | $\text{Ti}(\text{NO}_3)_4$ |
| 16) | diselenium diiodide | Se_2I_2 |
| 17) | copper (I) phosphate | Cu_3PO_4 |
| 18) | gallium oxide | Ga_2O_3 |
| 19) | tetrasulfur dinitride | S_4N_2 |
| 20) | phosphorus | PF_7 |